Organocatalysis

Reoptimization of the Organocatalyzed Double Aldol Domino Process to a Key Enal Intermediate and Its Application to the Total Synthesis of Δ^{12}-Prostaglandin J₃

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Abstract: Re-investigation of the L-proline catalyzed double aldol cascade dimerization of succinaldehyde for the synthesis of a key bicyclic enal intermediate, pertinent in the field of stereoselective prostaglandin synthesis, is reported. The yield of this process has been more than doubled, from 14% to a 29% isolated yield on a multi-gram scale (32% NMR yield), through conducting a detailed study of the reaction solvent, temperature, and concentration, as well as a catalyst screen. The synthetic utility of this enal intermediate has been further demonstrated through the total synthesis of Δ^{12}-prostaglandin J₃, a compound with known anti-leukemic properties.

Interest in the chemistry and biology of prostanoids continues unabated, with as many publications (ca. 600 per year) in the last decade as during their heyday of the 1970s.[1] Prostaglandins are an important class of potent lipid mediators that are involved in the regulation of many biological processes such as inflammation, pain response, and fever.[2] Consequently, this class of compounds has found wide-spread use as pharmaceuticals for the treatment of several diseases including pulmonary arterial hypertension[3] and glaucoma.[4] We recently described a dramatically short route to the prostaglandins, completing the total synthesis of PGF₁₀, in just seven steps (Scheme 1),[5] and its subsequent application to the concise synthesis of the related pharmaceutical analogues latanaprost and bimatoprost.[6] The key step in our synthesis is the double aldol dimerization of succinaldehyde with proline and dibenzylammonium trifluoroacetate (DBA) as catalysts to give the bicyclic enal intermediate (1) in high enantioselectivity (Scheme 1).[7] Whilst this single step converts a simple starting material into a complex intermediate fully primed to enable rapid attachment of the two side chains required, its Achilles’ heel is its low yield (caused by the extensive oligomerization of succinaldehyde under the reaction conditions).

In this paper, we describe a full re-investigation of this key step, which has culminated in an increase in yield from 14%[8] to a 29% isolated yield on a multi-gram scale, thereby enabling the enal 1 to be used not just in further, more efficient prostaglandin syntheses, but also as a highly functionalized and useful building block more generally. Furthermore, we demonstrate its application to the concise synthesis of Δ^{12}-PGJ₃, a prostaglandin of considerable contemporary interest due to its high activity against cancer stem cells, a class of cells that are notoriously resistant toward conventional chemical therapies.[4]

The aldol dimerization of succinaldehyde is complex: L-proline (2%) is added to a 2 M solution of succinaldehyde in MeTHF and after 24 h the mixture is diluted to 1 M and dibenzylammonium trifluoroacetate (5, 2%) is added. The two catalysts perform different roles: L-proline catalyzes the first aldol but does not catalyze the second aldol and 5 catalyzes the second aldol condensation and does not (and must not) catalyze the first aldol otherwise the reaction would occur with low enan-

Scheme 1. Previously reported L-proline catalyzed double aldol domino reaction for the synthesis of a key bicyclic enal intermediate 1 and its application to the total synthesis of PGF₁₀ (2).
tioselectivity. The catalysts must be added sequentially because 5 inhibits the first aldol reaction with proline and the time of addition is important since reaction of succinaldehyde with l-proline leads to oligomers over time. Under these conditions, a 14% yield was achieved (Table 1, entry 1), and so we embarked on a further optimization program re-investigating all the parameters. A small improvement in yield (16%) was achieved by changing the solvent to acetonitrile (Table 1, entry 2). The yield was further increased to 19% (NMR) by reducing the initial concentration to 1 mM (entry 3), but under these reaction conditions, the enol was isolated only in 9% yield. This disparity between the NMR and isolated yields was a consequence of the formation of a hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process. An extensive screening of the second catalyst was undertaken, and we found that the formation of the hemiaminal ether by-product 4, which formed upon addition of catalyst 5 to the lactol moiety of 1 during the isolation process.

Table 1. Re-optimization of the synthesis of enal 1.

<table>
<thead>
<tr>
<th>Entry</th>
<th>Solvent</th>
<th>2nd Cat.</th>
<th>Conc. 1 [%]</th>
<th>Conc. 2 [%]</th>
<th>Yield [%]</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>THF</td>
<td>RT</td>
<td>1.0</td>
<td>1.0</td>
<td>14</td>
</tr>
<tr>
<td>2</td>
<td>MeCN</td>
<td>RT</td>
<td>1.0</td>
<td>1.0</td>
<td>16</td>
</tr>
<tr>
<td>3</td>
<td>MeCN</td>
<td>RT</td>
<td>1.0</td>
<td>1.0</td>
<td>19 (9)</td>
</tr>
<tr>
<td>4</td>
<td>MeCN</td>
<td>RT</td>
<td>1.0</td>
<td>1.0</td>
<td>20 (20)</td>
</tr>
<tr>
<td>5</td>
<td>MeCN</td>
<td>65</td>
<td>2.0</td>
<td>2.0</td>
<td>23</td>
</tr>
<tr>
<td>6</td>
<td>EtoAc</td>
<td>65</td>
<td>2.0</td>
<td>2.0</td>
<td>21</td>
</tr>
<tr>
<td>7</td>
<td>EtoAc</td>
<td>65</td>
<td>0.75</td>
<td>0.75</td>
<td>33 (31)</td>
</tr>
<tr>
<td>8</td>
<td>EtoAc</td>
<td>65</td>
<td>0.75</td>
<td>0.75</td>
<td>35 (32)</td>
</tr>
</tbody>
</table>

[a] Reaction conditions (unless otherwise stated): Succinaldehyde (5.81 mmol), L-proline (2 mol%), solvent (X, M); then: 2nd catalyst (2 mol%). [b] Yield determined by 1H NMR spectroscopy using 1,3,5-trimethoxybenzene as an internal standard (isolated yields following chromatographic purification are shown in parentheses). [c] 5 mol% of the 2nd catalyst was used. [d] 40 hr reaction time for 1st step. [e] 50 g succinaldehyde (581 mmol) was used.

Evaporation followed by chromatography was ineffective, giving enal 1 contaminated by both oligomers and succinaldehyde. We were wary of carrying out an aqueous work-up because we had previously found that the enal is partially water soluble and required multiple extractions even from brine solutions, resulting in excessively large volumes of solvent. However, a recent report highlighting the beneficial effects of using Na₂SO₄ to salt-out water soluble compounds prompted us to reinvestigate aqueous work-ups. Using this strategy, just two extractions enabled full recovery of enal 1, which was purified by column chromatography; it was essential to pre-treat the silica gel with water (50 wt%) to ensure efficient removal of any remaining succinaldehyde and oligomeric material during chromatography. Using our optimized conditions and improved work-up and purification, 50 g of succinaldehyde was transformed into 12.8 g of enal 1 (29% isolated yield) as an inconsequential 4:1 mixture of diastereomers, and with the same enantiomeric ratio (99:1 er) as previously reported.

In order to further exemplify the synthetic utility of the enal 1, we chose Δ¹₂-prostaglandin J³ (Δ¹₂-PGJ³, Scheme 1) as a target because of its considerable biological potency towards stem cell cancer, nicotine and co-workers recently reported several strategies for the total syntheses of Δ¹₂-PGJ³. These strategies were subsequently applied to the development of a series of even more potent analogues as potential clinical candidates for stem cell cancer treatment, making this an even more sought-after target. We have previously used 1 to form the C12–C13 bond (prostaglandin numbering) in the syntheses of a variety of prostanoids through conjugate addition of a nucleophile to the electrophilic enol moiety (Scheme 2A). We reasoned that, by transforming enal 1 into enamidne 7, it would change the reactivity of the C12-position from electrophilic to nucleophilic, further broadening the utility of our key enal (Scheme 2A). In order to construct Δ¹₂-PGJ³, disconnection of the C12–C13 bond through an aldol/dehydration...
tion reaction sequence would require β-boryl aldehyde 8 and enone 9 (Scheme 2B). This type of aldol/dehydration strategy was originally reported by Kobayashi[16] and has subsequently been applied to the syntheses of Δ12-PGF2α[17] and other closely related prostaglandins.[19] Boryl aldehyde 8 was selected as a masked hydroxy aldehyde equivalent because it can be easily prepared by catalytic enantioselective conjugate borylation,[20] and subsequently unmasked later in the synthesis by stereospecific oxidation of the boronic ester.

Furthermore, it avoids potential elimination of a protected δ-alkoxy enone to the dieneone. Enone 9 could be obtained by olefination of hemiacetal 11, followed by hydrolysis of the carbamate group and elimination. The carbamate itself could be derived from carboxylic acid 13.

The synthesis of Δ12-PGF2α began through a double oxidation of enal 1 to the lactone-acid 13 using standard Pinnick oxidation conditions in 74% yield (Scheme 3). Carboxylic acid 13 was initially converted into acyl azide 14 in 80% yield. Heating 14 in toluene effected a Curtius rearrangement to afford an intermediate isocyanate, which was trapped with benzyl alcohol to give carbamate 12 in 90% yield. While it was possible to obtain 12 directly by adding benzyl alcohol to the cooled toluene solution, this led to sluggish nucleophilic capture reactions. It was crucial to remove toluene before alcohol addition to ensure a high yield of 12. Enecarbamate 12 was then reduced to hemiacetal 11 with Dibal-H in 98% yield. Initially, the Wittig reaction between hemiacetal 11 and phosphonium salt 10 was problematic due to poor conversion of 11 and the enamide product 15 was found to be difficult to isolate perhaps due to its limited stability. After considerable optimization, it was found that isolation of the enamide 15 could be avoided by first treating the hemiacetal 11 with excess phosphonium salt 10 and KOT-Amyl in THF. Direct addition of degassed water and para-toluenesulfonic acid to the reaction mixture effected the hydrolysis of the enamide group and dehydration to give enone 16 in 79% yield. The enone acid 16 was subsequently converted into the tert-butyl ester 9 in 83% yield.

The lower side chain was prepared in a three-step process from leaf alcohol 19 (Scheme 3). Following a known literature procedure,[21] 19 was converted into α,β-unsaturated ester 20 by initial oxidation to the corresponding alcohol, followed by a direct Wittig reaction in the same pot. Ester 20 was subjected to catalytic asymmetric conjugate addition with B2pin2[20a] to provide a β-boryl ester in high yield and enantiomeric ratio (85%, 93:7 e./r.). The boronic ester was subsequently reduced.
to the required β-boryl aldehyde 8 in 93% yield upon treatment with DIBAL-H.

To complete the synthesis of Δ12-PGJ3 (3), we commenced the development of an aldol reaction between enantiomerically enriched fragments 8 and 9 (Scheme 3). Initially, 1.2 equivalents of LDA were added dropwise to a solution of 8 and 9 at −78 °C to give the expected aldol product 17 as a mixture of C13-epimers in 23% NMR yield (Scheme 3). Increasing the loading of LDA to 2.0 equivalents dramatically improved the reaction to provide 17 in a 75% NMR yield. Due to the instability of β-hydroxy boronic ester 17 towards column chromatography, the crude reaction mixture from the aldol reaction was treated with MsCl and NET3 to give the corresponding mesylate, and subsequent elimination upon reaction with DBU produced exclusively the E-configured elimination product. The resulting boronic ester was oxidized to secondary alcohol 18 using NaBO3·H2O in 23% overall yield (over the three steps from 8 and 9). Finally, treatment with HBF4 gave Δ12-PGJ3 (3) in a 75% yield. The total synthesis of Δ12-PGJ3 (3) was achieved in 12 steps (longest linear sequence, LLS).

In conclusion, we have significantly improved the yield of our previously reported l-proline catalyzed double aldol dimerization of succinaldehyde from 14 to 29% for the synthesis of a key enal intermediate 1 that can be employed in the synthesis of a range of prostanoids. This has been achieved through a thorough re-evaluation of all of the reaction parameters, which led us to make four key modifications of the reaction conditions: changing Me-THF for EtOAc, changing dibenzylammonium trifluoroacetate 5 to thiomorpholinium trifluoroacetate 6, the temperature of the second step from 25 °C to 65 °C, and the concentration has been decreased from 2 M to 0.75 M in the first step and from 1 M to 0.35 M in the second step. The synthesis, and the practical isolation and purification of enal 1 on a decagram scale has also been developed. Furthermore, we have exemplified the synthetic versatility our enal intermediate 1 through its application to the total synthesis of Δ12-PGJ3 (3). This was achieved through an umpolung approach involving the conversion of the electrophilic enal moiety into an ene-carbamate, which serves as a masked nucleophilic moiety.

Acknowledgements

We thank EPSRC (EP/M012530/1) for support of this work. I.P.P. thanks AstraZeneca and the Bristol Chemical Synthesis Centre for Doctoral Training, funded by EPSRC (EP/G036764/1), N.S. thanks the Deutsche Forschungsgemeinschaft (DFG) and the Marie-Sklodowska-Curie Fellowship program (EC FP7 623426) for postdoctoral fellowships. A.P. thanks REGPOT-CT-2013-316149-InnovaBalt for partial fellowship support. We would like to thank Ian Davies (Princeton University) for useful discussions relating to the purification of enal 1.

Conflict of interest

The authors declare no conflict of interest.